## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS GCE Ordinary Level

## MARK SCHEME for the May/June 2008 question paper

## **5129 COMBINED SCIENCE**

5129/02

Paper 2 (Theory), maximum raw mark 100

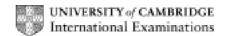
This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• CIE will not enter into discussions or correspondence in connection with these mark schemes.

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Page 2		Mark Scheme	Syllabus	Paper
		GCE O LEVEL – May/June 2008	5129	02
1	(a) (i) tree	e / grass / flower		[1]
	(ii) cow	v / snail / rabbit		[1]
	(b) Box 2 = Box 4 =			[2]
	(c) Sun / su	unlight (not light alone)		[1]
	(d) decomp	oser / bacteria / fungi		[1]
	energy i	supply is limited is used by the organisms any 2 is lost at each stage		
		ent energy left (for another level)		[2]
2	(a) copper	/ Cu		[1]
	(b) potassiu	um / K		[1]
	(c) iron / Fe	•		[1]
	(d) copper	/ Cu		[1]
	(e) zinc/Zr	ו		[1]
3		: It or 0.2 × 180 (0.6 gains 1 mark)		[2]
	(ii) V = = 1	IR or 7 × 0.2 .4		[2]
	<b>(b)</b> 0.6 / 2.0	)–(a)(ii)		[1]

	Pa	ge 3	Mark Scheme	Syllabus	Paper			
			GCE O LEVEL – May/June 2008	5129	02			
4	(a)	) limewater milky / cloudy / white (precipitate)						
	(b)		$H_4 = 16$ $D_2 = 44$		[2]			
		·	ightarrow 44 $ ightarrow$ 44 $ ightarrow$ 44 $ ightarrow$ 47 4 $ ightarrow$ 47 4 $ ightarrow$ 47 4 $ ightarrow$ 47 4 $ ightarrow$ 48 $ ightarrow$ 48 $ ightarrow$ 49 $ ightarrow$ 49 $ ightarrow$ 40 $ igh$		[2]			
5	(a)	blue	pink		[1]			
	(b)		nspiration		[1]			
		. , .	per surface has waxy layer fewer / no stomata swer could be in terms of lower surface		[2]			
	(c)	root ha			[2]			
6	(a)	reduct	on		[1]			
	(b)	conductile high do high m	any 2		[2]			
		-						
	(c)	boils a	t 100 °C/boils at single temperature		[1]			
7	(a)	(i) gr	avity / weight		[1]			
		(ii) gr	avitational / potential		[1]			
	(b)	line is	curved / not straight		[1]			
	(c)	F = ma = 3.75 m/s <sup>2</sup>	or a = F/m or 300/80		[3]			

	Page 4		Mark Scheme	Syllabus	Paper
			GCE O LEVEL – May/June 2008	5129	02
8	(a)	matt	black is a better absorber / shiny is a better reflector		[1]
	(b)	(i) s	stays the same / no change / none		[1]
		(ii) c	decreases / gets less / lowers		[1]
	(c)	micro	owaves and radiowaves (either order)		[2]
9	(a)	large mixes	grinds food pieces to smaller pieces s food with saliva elves (soluble particles) any 2		[2]
	(b)	secre lubric	ete liquid / saliva ete enzymes / amylase cate / softens food mes convert starch to maltose / sugar	ny 2	[2]
	(c)		eria les / enamel to dissolve / tooth decay		[2]
10	(a)	hydro	ogen / H <sup>+</sup>		[1]
	(b)	(i) r	red		[1]
		(ii) c	orange / yellow		[1]
	(c)	(i) N	$Mg + H_2SO_4 \rightarrow MgSO_4 + H_2$		[1]
		r	magnesium carbonate magnesium hydroxide any 2 magnesium oxide		[2]
11	(a)	like c	harges (repel)		[1]
	(b)	positi	ive		[1]

	Page 5		Mark	Scheme		Syllabus	Paper
			GCE O LEVEL	_ – May/June 2	800	5129	02
12		luency = I iod = s	Hz or s <sup>-1</sup>				[2]
13	(a)	and one	showing 3 bonding pairs lone pair shell drawn it must be col				[2]
	(b)	400–500 200–300 iron					[3]
	(c)	potassiu phospho	m rus (either order)				[2]
14	(a)	lack of (e	enough) food				[1]
	(b)	not enou too much civil unre earthqua	n rain / water / floods	arks	plants die / cr no photosynt plants washe no one to ten or crops dest food destroye	hesis / growth d away / die d crops royed	[4]
15	(a)	both pos	al shape with itive and negative values cycles shown	;			[3]
	(b)	increase stronger more tur			any 1		[1]
16	(a)	0.1 × 30 W = 0.1	= W × 0.2				[2]
	(b)		clockwise / iron rod goes attracted by the magnet	down / left goes	s down		[2]

	Page 6			Mark Scheme	Syllabus	Paper
				GCE O LEVEL – May/June 2008	5129	02
17	(a)	(i)	elec	tronic structure drawn as 2 8		[1]
		(ii)	+3			[1]
	(b)	on tl	ns a p he le	positive ion If the Periodic Table ross period is metal to non-metallic	ny 2	[2]
	(c)			d by a layer of (aluminium) oxide e layer / on surface of metal		[2]
18	(a)	C =		ledon nule		[4]
	(b)	wate oxyg suita	gen	/ named temperature		[3]
19	(a)			= 12 cm <sup>3</sup> rect reading from diagrams gains 1 mark)		[2]
	(b)	0.24	or	<b>(a)</b> /50		[1]
20	(a)	5				[1]
	(b)	exte load		n = 10 (cm)		[2]