UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS GCE Ordinary Level

MARK SCHEME for the May/June 2012 question paper for the guidance of teachers

5129 COMBINED SCIENCE

5129/02

Paper 2 (Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2012 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

	Pa	ge 2	Mark Scheme: Teachers' version	Syllabus	Paper
			GCE O LEVEL – May/June 2012	5129	02
1	(a)	(i) tr	ransport/carry oxygen ;		[1]
		n	resence of haemoglobin ; to nucleus present ; targe surface area (per volume)/biconcave disc shape ;	any 2	[max 2]
		n n	ombines with oxygen for transport/releases oxygen in nore haemoglobin contained within the cell/more oxyg- nore oxygen can pass into the cell (in lung capillaries), ell (in tissue capillaries);	en carried ;	leave the
		а	daptation and explanation must be linked correctly		[max 2]
	(b)	plasm	<u>na</u> ;		[1]
2	(a)		na or a = F/m or 0.32/0.2 ;		
		1.6 ; m/s²	(unit independent);		[3]
	(b)	2;			[1]
3	(a)	71; 7.1; 2.925	117; 11.7; (divided by 10) g (divided by 4) (ecf throughout)		[4]
	(b)	ionic/	electrovalent ;		[1]
	(c)	<u>kill</u> ba	acteria/micro-organisms/germs;		[1]
4	(a)	Nm ;			[1]
	(b)		applied further from fulcrum (pivot)/perpendicular distant er force gives same moment/larger moment for same		[2]
5	(a)	1 (r∈	elative charge) ; elative mass) ; elative charge) ;		[3]
	(b)	numb	er of neutrons/number of nucleons/mass number;		[1]
	(c)	same	number of electrons in outer shell ;;		[2]

Mark Scheme: Teachers' version

Syllabus

Paper

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6	(a)	(i)	B, C	c or D ;					[1]
		(ii)	<u>E</u> ;						[1]
	(b)	-	produces (hydrochloric) acid; kills bacteria (on food)/prevents food poisoning;						
		or acc	cept s	tores food ;	·				
			•	to eat constantly/car	n concen	trate on other a	ctivities ;		[max 2]
	(0)	hila		ld not be added (to th	no food b	oina diaceted) :			[4]
	(6)	the	fat (ii	ld not be added (to the the food) not emuls	sified;	,) .		[1]
		sto	mach	stion would be incomp a acid would not be ne	eutralised	l ;	any 2		
		act	ion of	f pancreatic enzymes	impaired	d ;	J		[max 2]
7	(a)	aeı	robic ((respiration) uses oxy	/gen, ana	erobic does no	:; ገ		
		aeı	robic ((respiration) releases ic (respiration) produ	more en	ergy than anae	robic ;	any 2	
				produces carbon diox			J		[max 2]
	(h)	bro	athin	g becomes more rapi	id/factor				
	(D)			g becomes deeper/la			st;		[2]
	, ,	<i>(</i> 1)	450	O()					F41
	(C)	(i)		<u>0</u> (m) ;					[1]
		(ii)		longer the distance the cept converse or corre			obic respiratio	n ;;	[2]
8	(a)			amplitude ; vavelength ;					[2]
				· · · · · · · · · · · · · · · · · · ·					[-]
	(b)			$\lambda = f/v \text{ or } 0.5 \times 6;$					[0]
		3.0) ;						[2]
9	(a)	cop	oper		₹	reacts vigorou	sly with stean	າ ;	
		ma	ignesi	ium	*	reacts vigorou	sly with water	•	
		iro	n	\rightarrow		no reaction ;			
			tassiu	m		reacts slowly v	vith cold wate	r and steam ·	[4]
		,00		/	-	Table comy	III. Joid Hato	. 2 otoum ,	[1]
	(b)	<u>lig</u> l	nted/k	ourning splint explode	es with a	pop;			[2]

	Page 4		Syllabus 5129	Paper 02
10	root hair osmosis xylem; transpira	;	5129	[4]
11		is positively charged ; ne charges repel ;		[2]
	(b) curr	ent;		[1]
12		I = P/V or P = VI or 60/240; = 0.25;		[2]
	(11)	E = Pt or P = E/t or E = VIt or 60 × 600 ; = 36 000 ; (600 max 1 mark)		[2]
	(b) (i)	microwave/radio;		[1]
	(ii)	X-rays/gamma rays ;		[1]
13	(a) met	hane ;		[1]
	(b) con	npound of carbon and hydrogen only ;;		[2]
	(c) 13	8 10 (all three);		[1]
	(d) (i)	sulfur dioxide ;		[1]
	(ii)	acid rain; corrodes buildings/kills plant or aquatic life;		[2]

Page 5)	Mark Scheme: Teachers' version	Syllabus	Paper
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14	(a)	(i)	<u>Sun</u>	• •		[1]
		(ii)	<u>light</u>	;		[1]
	<i>(</i> 1.)	(1)	4 /1			
	(b)	(1)		erbivores) ; arnivores) ;		[2]
		(ii)	6 (sp	pecies);		[1]
		(iii)	exar	gy lost at each trophic level; nple of energy loss (respiration, heat, digestion etc.) onger the food chain, the less energy (there is to pa	ss on);	any 2 [max 2]
	(c)			er population would decline/fall/less spiders ; eat moths/less food for spiders ;		
		mo	_	usshoppers/grasshopper population increases; wering plants/food for grasshoppers;		[max 2]
15	(a)	v = = 8		r 400/50 ;		[2]
	(b)	dire	ection	keeps changing/velocity is directional;		[1]
16	(a)	cop	per;			[2]
	(b)	(i)	cutle	ery/chemical plant/surgical equipment/named exan	nples ;	[1]
		(ii)		ng metals /adding other elements to a metal to chaples of changing property;	nange/improve	e properties ; [2]
17	(a)	ene	ergy c	an be neither lost nor created ;		[1]
	(b)	the	mical rmal/ etic ;	; heat ;		[3]

	Page 6		Mark Scheme: Teachers' version	Syllabus	Paper
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18	(a)	calcium o	carbonate ;		[1]
	(b)	aluminiu	m oxide ;		[1]
	(c)	potassiu	m nitrate ;		[1]
	(d)	calcium o	carbonate ;		[1]
19	(a)	iron (core	e) ; and secondary (correct way round) ;		[2]
	(b)	-	ging current/changing magnetic field (in iron core) ed e.m.f./voltage (in secondary)/current;	;	[2]
20		taken ; count rate	/activity/number of nuclei to halve ;		[2]