## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

**International General Certificate of Secondary Education** 

## MARK SCHEME for the October/November 2010 question paper for the guidance of teachers

## 0653 COMBINED SCIENCE

0653/23

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

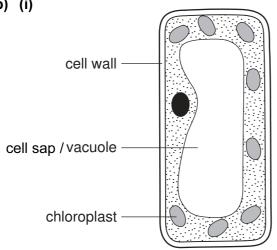
• CIE will not enter into discussions or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the October/November 2010 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

Page 2		2	Mark Scheme: Teachers' version	Syllabus	Paper
			IGCSE – October/November 2010	0653	23
1	(a) 1. 2.		s (layer of) air ; as insulator / reduces conduction / reduces convect	ion ;	[2]
	(b) (i)	incre	eased risk of skin cancer/eye damage/sun burn;		[1]
	(ii)		wave ; use ;		[2]
					[Total: 5]
2	(a) (i)		I oxide + carbon $\rightarrow$ lead + carbon dioxide S; RHS;		[2]
	(ii)	2.	lead oxide / carbon dioxide; ecf compounds contain more than one type of element reference to (different) elements / atoms in compound		bonded ; [3]
	(b) (i)	(dc)	power supply / battery / cell;		[1]
	(ii)	2.	chlorine; anode, non-metals form at the anode/chlorine is negative (and anode is positive);	s a non-metal/ch	nloride ions are [2]
					[Total: 8]
3	(a) (i)	tran	spiration / evapotranspiration / diffusion ;		[1]
	(ii)	ston	nata ;		[1]
	(iii)	2. 3.	condensation; water vapour cooled / temperature fell; gas changed to, liquid / water; ref. to particles and (kinetic) energy;		[max 2]

Page 3	Mark Scheme: Teachers' version	Syllabus	Paper
	IGCSE – October/November 2010	0653	23

(b) (i)



[max 2]

(ii) palisade (mesophyll);

1

[2]

- (iii) 1. carbon dioxide + water;
  - 2. glucose / carbohydrate / starch / sugar + oxygen;

[Total: 9]

- (a) 28(s); [1]
  - **(b) (i)** 1. distance = speed × time;  $= 10 \times 60 \times 9 = 5400 \,\mathrm{m}$ ; [2]
    - (ii) 1. work done = force × distance; 2. =  $10\ 000 \times 5400 = 54\ 000\ 000\ J$ ; [2]

[Total: 5]

- 5 (a) idea of restoring full number of chromosomes in the zygote; [1]
  - [1] **(b) (i)** ovary;
    - [1] (ii) oviduct / Fallopian tube;
  - (c) 1. produces / contains, (amniotic) fluid;
    - [2] protects/supports, embryo;
  - (d) 1. idea that mother's body needs to make substances for both herself and the fetus;
    - 2. iron for haemoglobin;
    - 3. calcium for, bones / teeth; [3]

[Total: 8]

		IGCSE – Octobe	er/November 2010	065	3 23	
(a) w	ater co	nducts electricity;			[1	
а	II symbo					
		e correct for 2 marks orrect for 1 mark			[3	
(c) (i	i) <b>K</b> ar	ıd <b>L</b> ;			[1	
(ii		J lights up ; K & L go off ;			[2	
( <b>d)</b> a	(d) add one 12 $\Omega$ to one 8 $\Omega$ (in series) ;					
( <b>e</b> ) <b>B</b>	(e) B F C D E A B F;					
D					[2	
					[Total: 10	
(a) (i	i) <b>O</b> ar	nd <b>S</b> ;			[′	
(ii	i)					
		element name	protons	neutrons		
		(oxygen)	8	8		
		nhaanharus	(15)	(16)		
		phosphorus	(10)	(10)		
	one	mark for each row ;;	(10)	(10)	[2	
(b) (i		<u> </u>		(10)		
(b) (i	i) copr i) mag	mark for each row ;;		(10)	[1	
	i) copr i) mag (allo	mark for each row ;;  per oxide / copper carbon  nesium ;		(10)	[1	
(ii	i) copp i) mag (allo i) reac com	mark for each row ;;  per oxide / copper carbon  nesium ; w aluminium)		(10)	[1 [1	
(ii	i) copp i) mag (allo reac com reac poly	mark for each row ;;  per oxide / copper carbon nesium ; w aluminium)  tion 1 bustion / oxidation ;	nate / other correct ;	(10)	[2 [1 [2 [2	

Mark Scheme: Teachers' version

Syllabus

Paper

Page 4

Page 5	Page 5 Mark Scheme: Teachers' version		Paper
	IGCSE – October/November 2010	0653	23

**8** (a) (i) 2000 (kg per hectare);

[1]

(ii) any two values with a range between 6 and 7.25;

[1]

- (iii) 1. calcium carbonate is a base;
  - 2. neutralise (acid in the soil);
  - 3. raise pH;
  - 4. above 5.5 / closer to 6.0;

[2 max]

- **(b)** 1. terracing/walls (qualified);
  - 2. bunds/embankment/ditch;
  - 3. plough along slope (not up and down);
  - 4. keep crop cover (at all times)/plant trees;
  - 5. other valid points;;

[2 max]

(c) (i) insects; allow self-pollination

[1]

- (ii) 1. pollen contains male gamete;
  - 2. fertilisation must take place;
  - 3. male gamete (in pollen grain) fuses with female gamete;
  - 4. seed develops from (fertilised) ovule;

[2 max]

- (iii) 1. biuret test/add biuret reagent/add copper sulfate and KOH solution;
  - 2. purple/lilac/mauve;

[2]

[Total: 11]

9 (a)

	description	charge	range in air	ionising ability
alpha	helium nucleus	positive	5 cm	very strong
beta	electron	negative	50 cm	medium
gamma	wave	none	many kilometres	weak

;;;; [4]

(b) alpha particles have low penetration in air/ will not reach people living in house/ smoke detectors are a long way from people;

[1]

[Total: 5]

Page 6	Page 6 Mark Scheme: Teachers' version		Paper
	IGCSE – October/November 2010	0653	23

- 10 (a) 1. adding chlorine;
  - 2. kills (harmful) bacteria/microorganisms/germs;
  - filtration :
  - 4. removes solid / insoluble materials / dirt;

4

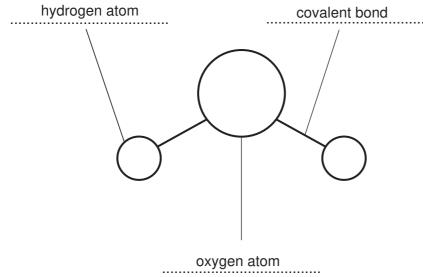
- (b) (i) 1. dissolves in/reacts with rain water;
  - 2. produces acidic solution / acid rain;
  - 3. ref. to sulfurous / sulfuric acid;
  - 4. acidic rain collects in rivers / lakes;
  - 5. reference to harmful effects of acidity, e.g. kills organisms;

max 3

(ii) removal of sulfur compounds from fuel / removal of sulfur dioxide from waste gases / reduce demand for energy / burn less fuel;

max 1

(c)



3

[Total: 11]