MARK SCHEME for the October/November 2011 question paper

for the guidance of teachers

0653 COMBINED SCIENCE

0653/23

Paper 2 (Core Theory), maximum raw mark 80

MMM. Hiremepapers.com

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2011 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.



	Page 2		Mark Scheme: Teachers' version	Syllabus	Paper	
			IGCSE – October/November 2011	0653	23	
1	(a) (i)	carb	on dioxide ;		[1]	
	(ii)	HC1	;		[1]	
	(iii)		tion has stopped ; has been used up/owtte ;		[2]	
	(iv)	(iv) calcium ;				
	(b) (i)	 (b) (i) carbon dioxide reacts with (sea)water ; makes water more acidic/less alkaline/pH decreases ; non-metal oxides are acidic ; 				
	(ii)	e.g. more surv	ept any reasonable science based idea: calcium carbonate may react with more acidic wat e difficult for coral to extract ions from sea/cora ive in more acidic water/enzymes (in coral) der ning;	al organism does	not	
		wan	, , , , , , , , , , , , , , , , , , ,			
					[Total: 8]	
2	(a) (i)	air re	esistance / friction / drag ;		[1]	
	(ii)	equa	al and opposite / cancel each other out ;		[1]	
	(iii)		stant speed ; w constant velocity)		[1]	
			= speed × time ; 600 = 288 000 m ;		[2]	
	(c) (i)	caus kills	ations / damage DNA ; se cancer ; cells ;			
			ation burns ; ation sickness ;		[max 2]	
	(ii)	(gra	nite) rocks ;		[1]	
	(d) name ;				101	
	ар	propria	ate use ;		[2]	
					[Total: 10]	

	Page 3		Mark Scheme: Teachers' version	Syllabus	Paper
			IGCSE – October/November 2011	0653	23
3	(a) glu wat		on dioxide ;		[2]
	ref.	ne blood / to haemo <u>ed</u> blood ([max 2]
	(c) (i)	0.4 dm ³	;		[1]
	(ii)	more (o	e answer refers to fast run unless otherwise stat xygen used per minute) ; es more rapidly ; more ;	ed)	[max 2]
	(iii)	muscles	<u>ergy</u> used when running faster ; <u>s</u> working harder ; e more <u>respiration</u> ;		[max 2]
	(d) bre	akdown c	of walls of alveoli / reduction of surface area ;		[1] [Total: 10]
4	(a) (i)	switches	s 1 and 2/both ;		[1]
	(ii)		er in parallel and ammeter in series ; ng else unchanged ;		[2]
	(b) (i)	coal/oil	/gas;		[1]
	(ii)	to reduc	e energy losses ;		[1]
	(iii)	(5000/4	400 000 = 10 000 / Ns, so Ns =) 800 000 (turns) ;		[1]
	(iv)	voltage for safet	needs to be lower ; ty ;		[2]
					[Total: 8]

	Page 4		Mark Scheme: Teachers' version	Syllabus	Paper	
			IGCSE – October/November 2011	0653	23	
5			nct insects ; luce, pollen / male gametes / male sex cells ;		[2]]
	(ii)	ovul ovar	-		[2]]

(b)

statement	asexual reproduction	sexual reproduction
gametes are involved	×	\checkmark
new individuals are produced	✓	\checkmark
a zygote is produced	×	\checkmark
offspring are genetically identical	✓	×

one mark for each correct row

(do not allow for anything where it is not clear whether it is a tick or a cross)	[3]
---	-----

[Total: 7]

6 (a) (i)	89 (%) ;	[1]
(ii)	metals are <u>melted</u> together ;	[1]
(iii)	iron ;	[1]
(iv)	unreactive ; strong / hard / not easily bent or deformed ; malleable ;	[max 2]
(b) (i)	tin oxide + carbon \rightarrow tin + carbon monoxide ;	[1]
(ii)	carbon ; gains / bonds with oxygen ;	[2]
(c) (i)	negative electrode ; compound in liquid form / solution / molten ; which conducts a current / contains free ions ;	[3]
(ii)	group number = outer electrons / Al is in Group 3 ;	[1]
		[Total: 12]

	Page 5		Mark Scheme: Teachers' version	Syllabus	Paper
			IGCSE – October/November 2011	0653	23
7	\ /		orce × distance ; 2000 = 200 000 J ;		[2]
	(b) (i)	kine	tic/movement ;		[1]
	(ii)	heat	:/sound;		[1]
	(iii)	surro	oundings ;		[1]
	(c) (i)	40 k	g;		[1]
	(ii)	volu = 40	me = mass / density ;)/1020 = 0.04 m ³ ;		[2]
					[Total: 8]
8	(a) (i)	dige	stion ;		[1]
	(ii)	so, r	nutrients / molecules, can be <u>absorbed</u> ;		[1]
	(iii)	-	eins ; gen ;		
			atured ;		[3]
	(b) (i)	the r	number of different, species / types of organisms ;		[1]
	(ii)		ct, food chains / food webs ; lators of frogs may reduce in numbers ;		
			cts / prey of frogs, may increase in numbers ;		[max 2]
					[Total: 8]

Page 6	6	Mark Scheme: Teachers' version	Syllabus	Paper
		IGCSE – October/November 2011 0653		23
9 (a) (i)		ied as fossil fuel / decay of organic matter / digestiv mants / vulcanism ;	ve system of	[1]
(ii)		nly other symbol ; 4 bonded to central C with all single bonds ;		[2]
(iii)	glob deta	bon dioxide) al warming / (runaway) greenhouse effect ; il of mechanism e.g. reflects heat back to Earth ; sing climate change / or example of ;		[max 2]
	(carl	bon monoxide) c (to humans) ;		[max 1]
(b) (i)	fract	ional distillation / fractionation ;		[1]
(ii)	the h	greater the molecular mass ; nigher the boiling point ; of the data e.g. C ₁₂ H ₂₆ most massive and has high	nest boiling point ;	[max 2]
				[Total: 9]