www.xiremepabers.com

UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the October/November 2011 question paper for the guidance of teachers

0653 COMBINED SCIENCE

0653/32

Paper 3 (Extended Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2011 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

1	(a)	(i)	carbon dioxide ;	[1]		
		(ii)	(there is not enough evidence) result shows that: carbonate present; but not calcium/need to show it is calcium carbonate;	[2]		
	(b)	(i)	carbon dioxide dissolves in/reacts with (sea)water/rain; makes water more acidic/less alkaline; non-metal oxides are acidic;	[max 2]		
		(ii)	accept any reasonable attempt at a scientific answer: e.g. calcium carbonate may react with more acidic water/lower pH makes it more difficult for coral to extract ions from sea/coral (polyps) does not survive in more acidic water;	[1]		
				[Total: 6]		
2	(a)	_	cose + oxygen → carbon dioxide + water ; marks for all correct, one mark if any mistake)	[2]		
	(b)		in the blood/in an artery/in a capillary; combined with haemoglobin/as oxyhaemoglobin; in red blood cells;			
	(c)	(i)	evaporation ; (evaporation) takes heat from body ;	[2]		
		(ii)	(assume answer refers to not drinking fluid unless otherwise stated) rose higher; rose faster;			
			use of comparative figures, e.g. 40.0 °C and 38.7 °C;	[max 2]		
		(iii)	less sweat produced when no fluids drunk/or reverse argument; to maintain water content of body/ref. to homeostasis;	[2]		
				[Total: 10]		
3	(a)		os layer of air ; is a good) insulator ;	[2]		
	(b)	ide	es not deplete fossil fuel reserves/non-renewable; a that dung is carbon neutral/renewable; osene is a hydrocarbon fuel;	[max 2]		

Mark Scheme: Teachers' version IGCSE – October/November 2011

Page 2

Syllabus 0653 Paper 32

	Page 3		3	Mark Scheme: Teachers' version	Syllabus	Paper
				IGCSE – October/November 2011	0653	32
	(c) (i) allow			v 20 – 100 Hz ;		[1]
		(ii) vibration passes ; from particle to particle ;				
		reference to rarefaction and compression/diagram; a series of (compressions and rarefactions)/diagram;				[max 2]
				[Total: 7]		
4	(a)	iror	۱;			[1]
	(b)	(i)	SnO	$_2$ + 2C \rightarrow Sn + 2CO ;; (symbols and balanced)		[2]
		(ii)	tin le A <i>l</i> m	ninium more reactive than carbon; ess reactive than carbon; eore strongly bonded to oxygen; w max 1 for the simple statement: aluminium is mor	re reactive)	[max 2]
		(iii)	alum alum ions	rence to use of carbon electrodes; hinium oxide is melted/dissolved in cryolite; hinium ions are positive/are cations; attracted move to negative electrode/cathode; gain electrons from/are discharged at negative ele	ctrode ;	[max 3]
	(c)	(i)		56 + 32 × 2/184 ; w 183.5)		[1]
		(ii)	7.80	\times 0.89 = 6.9(42)g (unit required);		[1]
						[Total: 10]
5	(a)	(i)		stigma ; anther/stamen ;		[2]
		(ii)	stam	na, feathery/outside flower/large surface area ; nen, dangling/outside flower ; etals ; (allow small petals)		[max 2]
		(iii)	(assı	ume answer refers to sexual reproducation unless over gametes;	otherwise stated)	[max 2]
			invol zygo	ves fertilisation ; te produced ; pring genetically different/not clones ;		[max 2]
				vhen nitrogen oxides, react with/dissolve in, (rain) v s plants ;	water ;	
			_	s aquatic animals ;		[max 3]
						[Total: 9]

	Page 4			Mark Scheme: Teachers' version	Syllabus	Paper
				IGCSE – October/November 2011	0653	32
6	(a) group of ce carrying ou			cells; out a particular or specific function/are similar cells	;	[2]
	(b)	(i)	•	eins ; no acids ;		[2]
		(ii)		sion ; o concentration gradient/from high concentration to	low concentration;	[2]
	(c)	(c) (assume answer refers to animal cells unless otherwise stated) look at cells (as opposed to whole organism); no cell walls; no large vacuoles;				
			•	oplasts ;		[max 2]
						[Total: 8]
7	(a)	(i)	swite	ch 1 and switch 2 ;		[1]
		(ii)	voltn	meter in parallel and ammeter in series;		[1]
	(b)	(i)	to re	educe energy losses ;		[1]
		(ii)	500	- Vs = Np ÷ Ns ; 0 ÷ 400 000 = 10 000 ÷ Ns / (Ns=) 800 000 (turns) ; nark for formula and 1 mark for substitution and answ	wer)	[2]
	(iii)		(in p	to alternating or changing voltage or current; orimary coil) produces alternating or changing magneric rence to alternating or changing magnetic field in co		
				ces (alternating) voltage in secondary coil; that size of voltage change depends on (ratio of) to	ırns :	[max 3]
				3 ()	- ,	_
						[Total: 8]
8	(a)	(i)		ned as fossil fuel / decomposition of organic matter / ems of ruminants / sources related to volcanism;	from digestive	[1]
		(ii)		covalent bonds means four pairs of electrons ; rect bonding diagram alone gains both marks)		[2]

Pag	je 5	Mark Scheme: Teachers' Version	Syllabus	Paper
		IGCSE – October/November 2011	0653	32
(b)	(i)	fractional distillation/fractionation;		[1
(the larger/heavier/greater surface area of/greater nu	mber of atoms in	
		molecules; the higher the boiling point;		
		OR unsaturation;		
		lowers boiling point (for similar molecular size);		[max 2]
(i	•	(shake liquid with) bromine/potassium manganate(V) mixture goes colourless if liquid is D ;	 ; 	
		because D is unsaturated ; (or reverse argument for A)		[3]
		(er reverse all gament ter 1 s)		[Total: 9]
9 (a)	(forc	ce =) mass × acceleration ;		
` ,	àcce	eleration = 1 200 000 / 400 000 ; m/s ² ;		[3]
	01	m/3 ,		[0]
(b)	(i)	to stop potato snacks oxidizing/reacting;		[1]
((ii)	pressure inside packet is greater than airplane pressu	re;	[1]
(c)	(i)	speed has magnitude only/velocity has magnitude ar	d direction ;	[1]
((ii)	A to B / C to D;		[1]
(i	iii)	(no) not a straight line;		[1]
(i	iv)	c ;		[1]
((v)	50 m/s;		[1]
()	•	the faster the skydiver travels the greater the air resis		
		eventually the air resistance balances the gravitational	I force ;	[2]
(v	/ii)	parachute increases air resistance;		[1]
				[Total: 13]

Mark Scheme: Teachers' version

Page 5

Syllabus

Paper