MARK SCHEME for the May/June 2014 series

0653 COMBINED SCIENCE

0653/22

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2014 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.



	Page 2	2	Mark Scheme Syllabus					
		IGCSE – May/June 2014 0653						
1	(a) (i)	5(m	ı);		[1]			
	(ii)	(ii) (speed =) distance/time ; = 5/0.2/= 25 m/s ;						
	(iii)		$n/s = 25 \times 3600 = 90000 m/hour$; 00 m/hour = 90 km/hour (which breaks speed limit)	of 80 km/h) ;	[2]			
	(b) (i)) (i) arrow pointing backwards, from van, labelled ' <u>friction (</u> from brakes)'; arrow pointing backwards, from van, labelled 'air resistance' (owtte); allow friction between tyre and the road						
	(ii)	conv	verted to <u>heat</u> ;		[1]			
					[Total: 8]			
2	(a) oxy	/gen; ter;			[2]			
	wa	,			[2]			
	(b) aid	s buo	yancy/helps it to float ;		[1]			
			aseous exchange ; rface would be under water ;		[2]			
	• •		s increase surface area/enable more water to be al led as roots immersed in/surrounded by water ;	osorbed/collected;	[2]			
	noi	neec	ied as roots initielsed in surrounded by water,		[2]			
			(from sewage);					
		-	∕gen in pond ; xygen suffocated fish / fish died through lack of oxyg	len ;				
	Sel	wage	is toxic ;		[max 3]			
					[Total: 10]			
2	(c) (i)	aalt	: A water		[4]			
3	(a) (i)		; A water		[1]			
	(ii)	hydı	rochloric acid ;		[1]			
	(b) Cu	₂ O ; A	Λ Cu ₄ O ₂ ;		[1]			
	(c) (i)	oxyg wate	gen/air ; er ;		[2]			

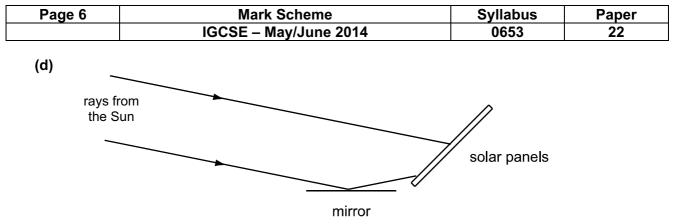
	Page		}	Mark Scheme	Syllabus	Paper	
				IGCSE – May/June 2014	0653	22	
		(ii)		t/oil/plating/any other correct method ; A alloy cclude oxygen/air/water/explanation matching met	hod ;	[2]	
						[Total: 7]	
4	(a)	car	bon d	ioxide and water ; (both needed, in any order)		[1]	
	(b)			asm ; embrane ;		[2]	
	(c)	larg into	ge/ins sma	own ; soluble molecules ; Il/soluble molecules ; n be absorbed ;		[max 3]	
	(d)	running ; 89 ; Allow ecf from mpt 1					
	(e)			kercise became more vigorous/faster/harder ; e (rate) increases ;		[2] [Total: 10]	
5	(a)	(i)	meta meta meta meta	al malleable, non-metal not malleable/ brittle ; al electrical conductor, non-metal insulator ; al heat conductor, non-metal insulator ; al ductile, non-metal not ductile ; al lustrous/shiny, non-metal not lustrous/dull ; al sonorous, non-metal not sonorous ; al high density, non-metal low density ;		[max 1]	
		(ii)	elen perio	nents become less metallic from Group 1 to 0/let	ft to right/across the	e [1]	
	(b)	(i)	hydr	um hydroxide ; rogen ; r order)		[2]	
		(ii)		d down the group is to lower melting point/other val d down the group is to more vigorous reaction ;	id response ;	[2]	

	Page 4		Mark Scheme	Syllabus	Paper
			IGCSE – May/June 2014	0653	22
	(c)	chlorine allow 2 n	atom loses (one) electron ; atom gains (one) electron ; narks for: ectron is transferred from the sodium atom to the chlo	orine atom	[2]
					[Total: 8]
6	(a)	colour ; platform anther ; pollen ; stigma ;		[5]	
	(b)	rough/so will stick		[max 1]	
					[Total: 6]
7	(a)				
		complete	e circuit, switch symbol, lamp symbol (all required);		[1]
	(b)	(i) 12(c	ohms) ;		[1]
		(ii) (curi	rent is) increased ;		[1]
		(iii) lamp	os are brighter/if one lamp goes out, the other is still	lit/owtte;	[1]
	(c)		d insulation/unsafe insulation ; hock/overheating/fire/other reasonable danger ;		[2]
	(d)		ed socket/current too high/overheating/gets too ho ts and breaks circuit ;	ot;	[2]
					[Total: 8]

	Page 5	5		Mark Scheme			Syllabus	Paper		
				IGCS	E – May/Jι	une 2014		0653	22	
8	(a) (i)	(i) pH increases ; contents become less acidic ;							[2]	
	(ii)	 (ii) carbon dioxide ; limewater ; cloudy/milky ; allow ecf from name of gas 								[3]
			-	gement for c aratus used		IS;				[2]
	(c) (i)	stee	per ini	tial gradient	with same	final volume	;			[1]
	(ii)	greater surface area ; greater rate of reaction/starts working more quickly ; [2							[2]	
	(iii)	 concentration of acid/pH of acid ; greater concentration/lower pH causes greater rate of reaction ; OR same size of tablet(s) ; 								01
		grea	ter sui	face area c	auses great	ter rate of re	action;		-	nax 2] al 12]
9	(a) (i)	cond	luctior	ı;						[1]
	(ii)	<u>conductor</u> of heat ;							[2]	
	(b) 05.	00 ; A	06.00)						[1]
	(c) (i)							[1]		
	(ii)			X-rays		visible light	infra-red	microwaves		

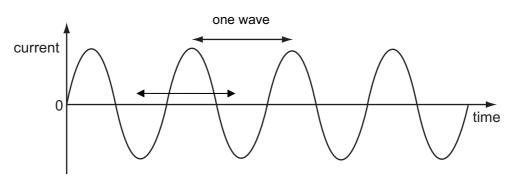
infra-red;
correct place in spectrum ;

[2]



direct ray + ray reflected from mirror to solar panel ; angle of incidence = angle of reflection as judged reasonable by eye ; [2]

(e) (i) one wave correctly marked and labelled (- example below);



(ii) 50 vibrations / oscillations (*allow* wavelengths) per second ;

[1]

[1]

[Total: 11]