## MARK SCHEME for the May/June 2014 series

## **0653 COMBINED SCIENCE**

0653/21

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

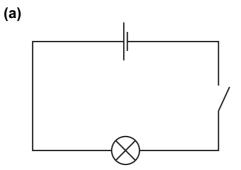
Cambridge is publishing the mark schemes for the May/June 2014 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.



Page 2			Mark Scheme	Syllabus	Paper
			IGCSE – May/June 2014	0653	21
(a)	(i)	hydr	rogen ;		[1]
	(ii)	flam	е;		
		pops			[2]
		(ect	for (a) (ii))		
	(iii)	mag	nesium		
		X			
		copp (i.e.	X below magnesium and above copper)		
		mag	nesium		
		X G			
			per;		
		(i.e.	G below magnesium and X in any order, and above	e copper)	[2]
	(iv)		/iron/A other metals with electronegativity betwee iron ;	en that of magnes	ium [1]
<i>(</i> L.)					
(b)	(1)		oval/loss of oxygen ; of electrons ;		[max 1]
		30			[
	(ii)	carb	on dioxide ;		[1]
(c)	(i)	P at	or near negative electrode within electrolyte;		[1]
	(ii)	bron	n <u>ine</u> ;		[1]
					[Total 10]

	Page 3				Mark Schem	e		S	yllabus		P	aper
				IGCS	∃ – May/Jun	e 2014			0653			21
2	(a)	Sun ;										[1]
	(b)	(i) oak	tree ;									[1]
		(ii) beet	les/gre	enfly/rabbi	ts/squirrels							[1]
	(c) oak tree → beetles → blackbirds → hawks ;; or oak tree → greenfly → frogs → hawks ;;											
		(1 mark o	correct	sequence o	f organisms,	1 mar	k correct a	arrows)				[2]
	(d)			ay decreas ay become	-							[2]
	(e)	(concent photosyr	,		; because	less	(carbon	dioxide	taken	in	for)	[2]
												[Total 9]

Page 4	Mark Scheme	Syllabus	Paper
	IGCSE – May/June 2014	0653	21

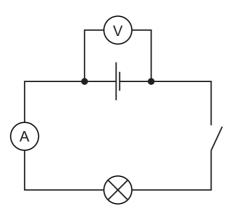


symbols all correct ; circuit connected correctly ; (either one or two cells used)

- (b) (i) (2) lamp needs (p.d. of) 3V (to light), so needs 2 × 1.5=3V cells (owtte); [1]
  - (ii) lamp takes <u>current of</u> 1.2A when lit (owtte) ;

(c)

3



voltmeter connected correctly ; ammeter connected correctly ;

[2]

[2]

[1]

[Total 6]

	Page 5			Mark Scheme	Syllabus	Paper
				IGCSE – May/June 2014	0653	21
4	(a)	(i)	fract	ional distillation/fractionation;		[1]
	<ul> <li>(ii) the lower the boiling point, the higher up the tower it is released/ condenses;</li> </ul>		released/	[1]		
	(iii) gasoline (petrol)/diesel/fuel oil/A kerosene ; used as <u>fue</u> l for transport/heating ;			[2]		
	(b)			gen: 78% ; jen: 21% ;		[2]
	(c)	<ul> <li>(c) (i) increase in water (vapour); increase in carbon dioxide; decrease in oxygen; temperature increases;</li> </ul>			[max 2]	
		(ii)		energy released/temperature increases ; substance(s) are formed ;		[2]
						[Total 10]

	Page 6			Mark Scheme	Syllabus	Paper
				IGCSE – May/June 2014	0653	21
5	<ul> <li>(right hand) no mark</li> <li>image laterally inverted (owtte);</li> </ul>					[1]
	(b) (i) electrical (energy) $\rightarrow$ sound (energy)			[1]		
	(ii) (frequencies lie) within human range 20 Hz to 20 000 Hz / (frequencies) are above 20 Hz and lower than 20 000 Hz ;			[1]		
	(c)	(i)	spee	ed = distance/time ; ed = 25/40 = 0.625/0.63 ; es/second/m/s ;		[3]
		(ii)	(100 (forc	N) es) are <u>equal</u> ;		[1]
		(iii)	one	complete wavelength correctly marked and labelled	;	[1]
		(iv)	amp	litude/frequency;		[1]

(d)

correct name ; correct box ;

[2]

[Total 11]

	Page		I	Mark Scheme	Syllabus	Paper
				IGCSE – May/June 2014	0653	21
6	(a)	(i)	zygo	ote/one of the ball of cells ;		[1]
		(ii) fertilization ;		ization ;		[1]
	(b)	<ul> <li>(b) to uterus/womb;</li> <li>(implants/embeds) in wall/lining of uterus;</li> </ul>			[2]	
	(c)	(i)		nin D A A/B/E/K ; ect use of named vitamin ;		[2]
	(d) 3.8 × 37; = 140.6/141;			[2]		
						[Total 8]
7	(a)			colour/gas to solid/increasing, mp/bp/density, density, density, density, density, density, density, density, d	own the group ;	[1]
	(b)	(i)	yella	ow/orange colouration ;		[1]
		(ii)		rine + potassium bromide $ ightarrow$ potassium chloride + b	promine	
		LHS ; RHS ;			[2]	
	(c)	covalent ;		[1]		
	(d)	makes water safe for consumption ; kills bacteria ;				[2]
						[Total 7]

	Page	e 8	Mark Scheme	Syllabus	Paper
			IGCSE – May/June 2014	0653	21
8	<b>(a)</b> a	air			
	v	vater <			
	n	netal /			
					[2]
	<b>(b) (</b> i		e energetic water molecules escape into air ; aining water has less (thermal) energy (so cooler)(	(owtte) ;	[2]
	<b>(</b> i	ii) cool	er water takes heat from air/water takes heat from	warmer air ;	[1]
	<b>(c)</b> a	allow spa	ace for (thermal) expansion ;		[1]
	(d) (i	i) 30×	: 15 × 10 = 4500 (cm <sup>3</sup> ) ;		[1]
	<b>(</b> i	<b>ii)</b> (der d =	nsity =) mass/volume/(d =) m/V ; 7500/4500 = 1.7/1.67 (g/cm <sup>3</sup> ) ;		<i>(ecf)</i> [2]

[Total 9]

Page 9	Mark Scheme	Syllabus	Paper
	IGCSE – May/June 2014	0653	21

9 (a)	
	diagram

diagram	name of cell	function of cell
	red blood cell	transport of oxygen;
	white blood cell	defence against disease / phagocytosis;

(b) right; pulmonary artery; [3] valves; [1] (c) (i) oxygen; (ii) glucose/sugar/amino acids/(any named) vitamin/(named) mineral/water/ [2] carbon dioxide ;;

[Total: 10]

[4]