

Cambridge International Examinations Cambridge International General Certificate of Secondary Education

COMBINED SCIENCE

0653/22 May/June 2016

Paper 2 Core Theory MARK SCHEME Maximum Mark: 80

Published

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International Examinations

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1 (a)

function	name of organ(s)
ingestion	mouth ;
absorption of digested food	small intestine ;
secrete digestive enzymes	salivary glands ; small intestine ; pancreas ; max 2

[4]

	(b) pla	sma ;	[1]
	(c) diff fro	usion ; m high concentration to low concentration ;	[2]
	(d) (i)	pH 2.7 allow 0.1 pH tolerance ;	[1]
	(ii)	activity would disappear ; graph shows no activity above pH 4.5 ;	[2]
2	(a) (i)	electrolysis ;	[1]
	(ii)	name: bromine ; colour: brown/orange-brown ;	[2]
	(b) coj	oper chloride \rightarrow copper + chlorine ;	[1]
	(c) (i)	increase;	[1]
	(ii)	electron ; proton ; neutron ;	[3]
	(iii)	no. protons + no. neutrons/number of particles in the nucleus ;	[1]
3	(a)	weight/gravitational (force) ; accept gravity	[1]
	(b) (i)	<i>Either</i> it does not affect the speed <i>(no mark)</i> weight/force/gravity acts downwards ;	
		<i>or</i> it decreases the speed of the cart <i>(no mark)</i> due to friction / frictional forces ;	[1]
	(ii)	(average) speed = distance/time (or rearranged) ; time = (distance/speed) = 20/8 = 2.5(s)	[2]

P	age	3	Mark Scheme	Syllabus	Paper
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		(iii)	horizontal straight line for constant speed/ slightly sloping line for decreasing speed ; smooth sloping line (straight or curved) down to speed = 0 ;		[2]
	(c)	•	om) potential (energy)/gravitational potential (energy) ; htermal/heat (energy) ;		[2]
4	(a)	ion xyle	l membrane ; s ; em ; nspiration ;		[4]
	(b)		a of: t hair cells are very delicate/fine/are easily damaged (by soil)/owtte	э;	[1]
	(c)		carbon dioxide + water ; (→) sugar/glucose + oxygen ;		[2]
		(ii)	light ; supply of carbon dioxide ; chlorophyll/chloroplasts ; (suitable temperature) ;		[max 2]
5	(a)	(i)	fractional distillation ;		[1]
		(ii)	(compound/molecule) containing hydrogen and carbon ; only ;		[2]
	(b)	(i)	methane;		[1]
		(ii)	oxygen ;		[1]
	(c)	(i)	C₂H₅ correct ; –O-H correct ;		[2]
		(ii)	carbon dioxide ; water/steam/water vapour ;		[2]
6	(a)	the	rmal expansion (of sea water) ; owtte		[1]
	(b)	(i)	evaporation ;		[1]
		(ii)	no effect ; decrease/cool ;		[2]

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(c)	(i)	radiation ;							[1
	(ii))							
		gamma rays	X-rays		(visible) light	infra	red		radio waves
	1. infrared ; – must be circled 2. (visible) light ; – must not be circled infrared in correct space ; light in correct space ;							[4	
(a)	(i)	organism	producer	consumer	n herbivo	ore (carnivo	ore	
		buzzard		✓			✓		
		grass	~						
		snail		~	~				
		thrush		~			✓		
		one mark for	each correct li	ne ;;;					[3
	(ii)	grass → snail organisms in arrows in corr	correct order ;						[2
(b)	(i)	keeping cattle	e/growing rice	/leaving rub	bish in dum	ips/avp	;		[1
	(ii)	it is a greenhouse gas/traps heat/infra-red radiation ; it contributes to global warming ;							[2

8 (a) (i) (most reactive) calcium zinc iron [1] copper; (ii) bubbles of gas/fizzing/effervescence/dissolving; [1] **(b)** (test) aqueous sodium hydroxide/aqueous ammonia; (iron(II) ions) (gelatinous) green precipitate/green solid; [3] (iron(III) ions) brown precipitate/brown solid; (c) (i) exothermic; [1] (ii) 1+/+1/Na⁺/Na¹⁺; [1] (iii) (sodium atom) loses one/an electron ; [1]

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9	(a) (i)	resistor ; accept variable resistance/rheostat		[1]
	(ii)	changes/varies current ; changes/p.d. across the buzzer;owtte		
		changes the resistance in the main circuit ;		[max 2]
	(iii)			
		ammeter symbol; ammeter in series with buzzer (any correct point in circuit, <i>reject</i> if in voltmeter branch); all else correct (ignore tiny gaps in wiring);	n the	[3]
		e of correct reading off graph at $6V > 0.015A$; istance at $6V = 6/0.015 = 400(\Omega)$;		[2]
	• •	quency unchanged/remains the same ; plitude increases ;		[2]