

## **Cambridge International Examinations**

Cambridge International General Certificate of Secondary Education

COMBINED SCIENCE 0653/61

Paper 6 Alternative to Practical

October/November 2016

MARK SCHEME
Maximum Mark: 60

## **Published**

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Page 2	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
1(a)	time and minutes;	2
	observation/appearance;	
1(b)	iodine (molecules) move into bag/tube ; by diffusion/because they are small enough ; starch and iodine produce blue-black/darker colour ;	3
1(c)(i)	(blue-black) colour goes/fades/goes brown;	1
1(c)(ii)	no starch left/no starch-iodine complex ;	1
1(c)(iii)	Benedict's solution; heat; yellow/green/orange/red;	3
	Total:	10

Question	Answers	Marks
2(a)(i)	add sodium hydroxide (solution)/NaOH ; green ppt. ;	2
2(a)(ii)	add dilute nitric acid/HNO $_3$ ; then add barium nitrate solution/Ba(NO $_3$ ) $_2$ ; white ppt.;	3
2(b)(i)	hydrogen/H <sub>2</sub> ;	1
2(b)(ii)	white ppt.; ppt. dissolves/becomes colourless solution/soluble in excess;	2

Page 3	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
2(c)(i)	displacement/redox/cation reduced/it is reduced/Fe <sup>2+</sup> goes to Fe/it is replaced by Mg/it is replaced by Mg <sup>2+</sup> ;	1
2(c)(ii)	exothermic;	1
	Total:	10

Question	Answers	Marks
3(a)(i)	6.5 ;	1
3(a)(ii)	65 ;	1
3(a)(iii)	appropriate precaution (either written or shown on diagram); take reading at eye level/use of set square to ensure rule vertical /use of fiducial aid;	max 1
3(b)	31. <u>0</u> ;	1
3(c)	$T = 1.55$ ; $T^2 = 2.4$ ;	2
3(d)	suitable choice of scales (more than half the grid used); at least 4 plots correct to ½ small square; good best-fit straight line with a ruler, omission of anomalous point;	3
3(e)	yes agree (no mark) (straight) line through the origin	max 1
	no disagree (no mark) all points / anomaly not on the (straight) line ;	
	Total:	10

Page 4	Mark Scheme	Syllabus	Paper
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Question		Answers				Marks	
4(a)(i)	dish	Seedling h	eight/mm		Average		2
		Student 1	Student 2	Student 3	seedling height/mm		
	Α	32	34 ;	30 ;	32		
	В	14	18	16	16		
	С	2	3	1	2		
4(a)(ii)	average	s calculated co	orrectly – A	= 32, B = 16	, C = 2 ;		1
4(b)(i)	linear ve	horizontal axis labelled dish A, B and C; linear vertical axis labelled height/mm and uses at least half the grid; all bars correct height ± one half small square;			3		
4(b)(ii)	reduces	effect of errors	s/takes acc	ount of varia	tions/anomal	lies accounted for/some results may be error ;	1
4(c)		it/shorter/stur oncentration h			;		2
4(d)	same sp	same specified environmental conditions, e.g. temperature, light (intensity);					max 1
						Total:	10

Page 5	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
5(a)	oxygen/O <sub>2</sub> ;	1
5(b)(i)	litmus; blue to red gas is acid <b>AND</b> red to blue gas is alkaline; OR UI/full range indictor/pH indicator; red/orange/yellow/any pH less than acid <b>AND</b> blue/purple/any pH greater than 7 alkali;	2
5(b)(ii)	gas will not change the colour of red and blue litmus/ UI and green or pH 7;	1
5(c)	diagram showing the inverted <b>test</b> -tube with the open end under water ; water risen into the test-tube ;	2
5(d)	gas $\mathbf{V}$ = ammonia/NH <sub>3</sub> ; gas $\mathbf{W}$ = hydrogen chloride/HC $l$ /sulfur dioxide/SO <sub>2</sub> ;	2
5(e)	add limewater to test-tube and shake ; (limewater goes) white precipitate/milky ;	2
	Total:	10

Page 6	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
6(a)	A in series with the power source <b>AND</b> V in parallel ;	1
6(b)	0.65 (A); 1.5 (V);	2
6(c)	wire $L = 1.5$ ; wire $M = 1.5/0.65 = 2.3$ ; ohms/ $\Omega$ ;	3
6(d)	minimum of 3 lengths; minimum 10 cm range; control <b>ONE</b> from material/cross-section/temperature; graph of resistance against length;	4
	Total:	10