## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

**International General Certificate of Secondary Education** 

## MARK SCHEME for the October/November 2006 question paper

## 0654 CO-ORDINATED SCIENCES

0654/02 Paper 2 (Core Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and students, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

The grade thresholds for various grades are published in the report on the examination for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses.

CIE will not enter into discussions or correspondence in connection with these mark schemes.

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	Page 2		Mark Scheme	Syllabus	Paper
			IGCSE - OCT/NOV 2006	0654	02
1	Hemiph Ninox no		gops habroptilus D; niphaga novaeseelandiae B; ox novaeseelandiae E; alea regia C;		[4]
	(b)	(i)	one word is its genus; other word is its species;		[2]
		(ii)	name is Latin and made up of two words;		[1]
				I	[Total: 7]
2	(a)	(i)	ammeter;		[1]
		(ii)	2 coulombs ;		[1]
		(iii)	R = V/I ; = 12/2 = 6 ohms ;		[2]
		(iv)	diagram to show clearly that the bulbs are in series;		[1]
		(v)	12 ohms ;		[1]
	(b)	(i)	in correct position to control motor and other switches etc	•	[1]
		(ii)	power = voltage x current; = 5 x 220 = 1100 W;		[2]
				,	[Total: 9]
3	(a)	(i)	rusting not expected in either tube; rusting requires air/oxygen and water (together); nail in A has no water; nail in B has no air/oxygen;		[max 3]
		(ii)	paint would be the barrier of choice; second mark for a reason why paint is suitable or why one not;	e or both of the others	s is <b>[2]</b>
	(b)	(i)	3;		[1]
		(ii)	chromite reduced since it loses oxygen; carbon oxidised since it gains oxygen;		
			or carbon oxidised and chromite reduced; reference to oxygen gain or loss;		[2]
				ı	[Total: 8]

Page 3			Mark Scheme	Syllabus	Paper
4 (a)	(i)	В	scapula ulna humerus ;	0654	02
		D	tendon;		[4]
	(ii)	line	e to space within elbow joint or shoulder joint;		[1]
	(iii)	lub	prication/reduce friction;		[1]
(b)	(i)	he	at/touching the hot object;		[1]
	(ii)	bic	reps muscle ;		[1]
	(iii)	alo	an electrical impulse ; ong a nerve/carried by nerve ; ong a motor nerve cell ;		[max 2]
	(iv)	rela	axes/is stretched ;		[1]
				["	Гotal: 11]
5 (a)		•	will be absorbed/will not pass through paper ; e/no gamma will be absorbed ;		[2]
(b)	(i)	11	0 130 150 all required for mark ;		[1]
	(ii)		unt is increasing ; thickness is decreasing ;		[2]
(c)	(i)	ph	monitor technician's exposure to radiation; otographic film is sensitive to radiation; edarker the film goes the greater the exposure;		[max 2]
	(ii)	fab	oric will absorb some radiation ;		[1]
(d)	uran	ium,	fission, heat, turbines, generators ;;;		[3]
(e)	fossi	il fue	els are a finite resource; (accept environmental answers)		[1]
				[	Total: 12]
6 (a)	(i)	_	oup of atoms/more than one atom ; nemically) bonded/joined ;		[2]
	(ii)	hyd	drogen ;		[1]
(b)	(i)	Y Z spo	alanine glycine lactic acid (all correct); ots for unknowns at the same position/height/travelled same own substances;	e distance as	[2]
	(ii)		w substances have been made/these are larger molecules sve changed/joined/other reasonable;	so smaller ones	[1]
	(iii)	pro	oteins/polypeptides ;		[1]
	(iv)	•	lymer is much larger/heavier/in the form of long chain/is mad elecules linked together ;	de of amino acid	[1]
			© UCLES 2006		[Total: 8]

	Pa	age 4		Mark Scheme	Syllabus	Paper
				IGCSE - OCT/NOV 2006	0654	02
7	(a)	(i)	labe	I to outer layer ;		[1]
		(ii)	no c	hloroplasts ;		[1]
	(b)	(i)		water ; s milky ;		[2]
		(ii)		iration ; east (cells) ;		
				ose combining with oxygen ;		[max 2]
						[Total: 6]
8	(a)	(i)	blue	and green ;		[1]
		(ii)	cyar	n;		[1]
		(iii)	refle	cted by fabric ;		[1]
	(b)	(i)		sity = mass/volume ; kg/dm³ ;		[2]
		(ii)	40(N	N);		[1]
	(c)	= 4	rk = F 0 000 x 10 x			[2]
	(d)	son only	ne mo y fast	eat causes particles to move faster; blecules will be moving faster than others; est molecules have enough energy to escape; ries away water particles;		[max 3]
	(e)	ligh	ıtweig	ht, waterproof, strong, rotproof, unreactive; ;		[2]
						[Total: 13]
9	(a)	coa	ı <u>l m</u>	ethane ;		[1]
	(b)		bon d ter ;	lioxide ;		[2]
	(c)			e to non-polluting emissions/water will not cause pollution; al detail e.g. reduced health risks from CO/particulates;		[2]
	(d)	(i)	an e	gnesium sulphate) lectrolyte contains dissolved ions/for cell to work the solution nesium sulphate is ionic/forms free ions when dissolved;	n must conduc	t ; <b>[2]</b>
		(ii)		<b>D</b> or <b>E</b> ; cell to work) electrodes must be dissimilar metals ;		[max 2]
						[Total: 9]

Mark Scheme

Syllabus

Paper

Page 5			Mark Scheme	Syllabus	Paper
			IGCSE - OCT/NOV 2006	0654	02
10 (a)	(i)	ciro	cle around a flower or the fruit ;		[1]
	(ii)	sqı	uare around one of the little plantlets ;		[1]
(b)	(i)	ova	ary ;		[1]
	(ii)	les	n colonise new areas ; ss competition with parent plant ; light/water/nutrients ;		[max 2]
	(iii)	oxy	ter ; ygen ; itable temperature ;		[3]
					[Total: 8]
11 (a)	(i)	2	ter only in both <b>2</b> and <b>3</b> ; spaced (three to five particles) ; random and close (at least eight particles);		[3]
	(ii)		d (acidified) silver nitrate (solution) ; ositive test for chloride ions is) white precipitate ;		[2]
(b)	(i)	rer	moves insoluble material/reasonable example of;		[1]
	(ii)	chl	orine/ozone ;		[1]
	(iii)		e/calcium carbonate/probably have to accept any correct ; cause water is acidic ;		[2]
					[Total: 9]