UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

MARK SCHEME for the May/June 2009 question paper

for the guidance of teachers

0654 CO-ORDINATED SCIENCES

0654/02

Paper 2 (Core Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

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	Page 2			Mark Scheme: Teachers' version	Syllabus	Paper
				IGCSE – May/June 2009	0654	02
1	(a)	(i)	5.1 ((±0.1) ;		[1]
		(ii)		adiation dose goes up so do incidences of leukaemi dences are (directly) proportional ;	a/radiation dose an	d [1]
	((iii)	radia	ation burns/radiation sickness/death/genetic mutatio	n ;	[1]
	(b)			stopped by paper etc ; ot stopped by paper/only partially stopped by alumin	ium	[2]
	(c)	(i)	time	taken for half atoms to decay/time taken for count/r	ate to decrease by	half ; [1]
		(ii)		ılf lives ; lays ;		[2]
						[Total: 8]
2	(a)	A B C D	cond trans	poration ; densation ; spiration/evaporation ; sipitation ;		[4]
	les les mo		s wate s raint re rur	•		[max 2]
	(c)	(i)	plas	ma ;		[1]
		(ii)	mov ppm from	nosis ; res/diffuses, through partially permeable membrane n. is cell membrane n where there is a lot of water to where there is less/ ential to low water potential ;		[max 3]
				teria/micro-organisms/pathogens ; ht cause illness/example of illness ;		[2]
						[Total: 12]

	Page 3		Mark Scheme: Teachers' version	Syllabus	Paper	
			IGCSE – May/June 2009	0654	02	
3		 (a) a group of atoms ; (covalently) bonded ; (b) the hard water/hardness in A reacted with the soap ; soap precipitated as scum ; less soap available to help washing/soap needed to improve washing ; 				
	SOS					
	(c) (i)	11;			[1]	
	(ii)	2 ; Ca ii	n Group II/20 electrons with e.c. 2.8.8.2/calcium ha	s valency of 2 ;	[2] [Total: 8]	
4	(a) (i)	varia	neter in parallel with lamp ; able resistor and ammeter in series ; ything else correct ;		[3]	
	(ii)	to va	ary current/voltage/potential difference (through/acro	oss lamp) ;	[1]	
	(iii)	R = = 5.3	V/I ; 3; (allow in working rather than in table)		[2]	
	(iv)	resis	nent gets hot ; stance is not constant ; age and current are not directly proportional ;		[max 2]	
	(b) dar shc	mageo ort circ		[2] [Total: 10]		
5	(a) (i)	feath beal wing			[max 2]	
	(ii)		ptiles ; nphibians ;		[2]	
	(iii)	Ran	a temporaria ;		[1]	
	(iv)		bed feet ; surface area) for pushing against water (when swin	nming) ;		
			s near top of head ; ee above water surface ;			
			ng hind legs ; ush against water (when swimming) ;		[2]	

Page 4			Syllabus	Paper
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(b)	(i)	${f A}$, because the body temperature does not change ;		[1]
((ii)	mammals ;		[1]
(i		good environment for cells ; enzymes have optimum temperature ; idea of affecting rate of (metabolic) reactions ; can be active in all temperatures ;		[max 2]
(i	•	food used to generate heat/keep warm ; in respiration ;		[2]
				[Total: 13]
6 (a) I	mag	gnesium sulphate ; + hydrogen ;		[2]
(b)	(i)	(Expt. 2) shortest time to collect 30 cm ³ gas/same volu	me of gas/OWTTE ;	[1]
(reduce (acid) temperature ; reduce acid concentration ; decrease surface area of magnesium/use same mass	of Mg but larger pieces	s; [max 2]
(i		reaction ; is exothermic ; releases (heat) energy ; which is transferred to the flask/surroundings ;		[max 2] [Total: 7]
7 (a)	•••	working; = 5000 N ;		[2]
((ii)	$15 - 25 \text{ N/cm}^2$; explanation e.g. uses 50 N/cm^2 at 40 m and 10 N/cm^2 a	at 0m ;	[2]
		mentum = m x v ; .2 x 10 = 12 kg m/s ;		[2]
(c)	(i)	any electromagnetic wave etc ;		[1]
(vibrations at right angles to direction of wave ; (transverse)		
		or vibrations in same direction as wave ; (longitudinal)		[1]
				[Total: 8]

	Pag	e 5	Mark Scheme: Teachers' version	Syllabus	Paper	
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8			ling C /no tip ; no tip) and did not grow ;		[2]	
		 (b) both have grown (taller); shoot B has bent towards the light but shoot A has grown straight up; 				
	• •	 (c) seedling D/tip covered ; (had its tip covered) and has not grown towards the light ; 				
		(d) for photosynthesis ; light is energy source ;				
		•	can grow faster with more light ;		[max 2]	
					[Total: 8]	
9	(a) ((i) ∈	electrode connected to negative side of power pack la	abelled ;	[1]	
	(i	ii) c	chlorine ;		[1]	
	(ii	ii) h	nydrogen ;		[1]	
	(iv	•	because solution becomes alkaline ; because sodium hydroxide is formed in the solution ;		[2]	
	(b) (halogen) displacement/redox ; chlorine is more reactive than iodine ;		[2]	
	(i	ii) c	chlorine + potassium iodide $ ightarrow$ potassium chloride + i	odine ;	[1]	
					[Total: 8]	
10	(a) (he current alternates/is alternating ;		101	
			50 times per second ;		[2]	
	(i	ii) c	current = 5000/250 = 20A ;		[1]	
	• •	(b) efficiency = useful energy out/total energy in ; half energy is wasted ;			[2]	
		 (c) aluminium is a good conductor of heat ; wood good insulator/heat cannot travel through (and burn hand); 				

Page 6	ge 6 Mark Scheme: Teachers' version Syllabus		Paper	
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11 (a) (i)	prote	eins/peptides/polypeptides ;		[1]
(ii)	Ν;			[1]
(b) (i)	b) (i) hexane has a lower boiling point/is more volatile/evaporates more easily ;			; [1]
(ii)		pound of only carbon and hydrogen ; h contains only single (covalent) bonds ;		[2]
(iii)	in pa	trons are shared ; airs/one electron from each atom is shared/OWTTE rence to full outer shell ;	;	[max 2]
(c) (i)	a "ca carb soyt	iesel is a renewable energy source ; arbon neutral" energy source ; on dioxide produced is removed from the peans/carbon dioxide is re-used/OWTTE ; ne process of photosynthesis ;	atmosphere by	growing (new) [max 2]
(ii)	acid thes	ir compounds burn to produce SO ₂ /sulfur oxide ; rain; e cause damage to buildings/irritate respiratory sys a cost involved in removing sulfur from diesel ;	stems ;	[max 2]
				[Total: 11]