UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

## MARK SCHEME for the October/November 2011 question paper

## for the guidance of teachers

## 0654 CO-ORDINATED SCIENCES

0654/22

Paper 2 (Core Theory), maximum raw mark 100

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Mark schemes must be read in conjunction with the question papers and the report on the examination.

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1 (a)

2

	insect		1			2		3		4	namo
			а	b	а	b	а	b	а	b	name
	Α		$\checkmark$			~					Musca
	В			~				~			Formica ;
	С			✓			~				Termes ;
	0	0	$\checkmark$		~				~		Graphosoma ;
	E	Ξ	$\checkmark$		✓					✓	Coccinella ;
(b)	<ul> <li>one mark per correct row (if no ticks, or if all ticks are wrong, allow one mark for getting all four names correct) [4]</li> <li>(b) idea of universal understanding/to identify its genus and species ; [1]</li> </ul>										
$(\mathbf{c})$	(i)	proto	ase/tr	uncin /r	oncin	0		pase ;			[1]
(C)	.,		o acids		Jepsin	0			la and	alvoorol	[1]
	(11)	amm		)		0			is anu i	glycerol	l; [1]
(d)	(i)	white	e (blood	l) cells	;						[1]
	<ul> <li>(ii) they can kill other organisms ; cause disruption to food chains ; they could harm human health ; pests can become resistant to the pesticides ;</li> <li>[max 1]</li> </ul>						[max 1]				
(e)	(i)		tions ; o air pa	rticles	/compr	essions	s and ra	arefacti	ons ;		[2]
	(ii)				er pitch quency						[2] [Total: 13]
	<i>(</i> <b>1</b> )										
(a)				-	awn an	id label	ied ;				[1]
		-	l and o		Э;						[1]
	(iii) constant s			ant speed ;							[1]

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(b)	(i)	cause cancer/kills cells/radiation burns/ra mutation/change DNA ;	[2]					
(	(ii)	(granite) rocks/radon ;	[1]					
$(\mathbf{a})$								
	use	tromagnetic wave ; ;		[2]				
(d)	(i)	to stop snacks spoiling/oxidising ; stops microbes respiring ;						
		nitrogen is unreactive ;		[max 2]				
(	(ii)	pressure inside packet is greater than airpl	ane pressure ;	[1]				
				[Total: 11]				
3 (a)	(i)	4;		[1]				
(	(ii)	silicon ; (does have same outer electrons) because	in same group ;	[2]				
(i	iii)	neon ;		[1]				
(b)	(i)	formed as fossil fuel ;						
		decomposition of organic matter ; from digestive systems of ruminants ;						
		ref. volcanism ;		[max 2]				
(	(ii)							
		exothermic means heat given off ; much heat per second means rate is high ;		[max 2]				
(c)	(i)	detergent/soap ;		[1]				
(	(ii)	suitable example of water pollution ;		[1]				
				[Total: 10]				

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4	• • •	(a) traps layer of air ; ref. good insulator ;						
	(b) (i)	(weight =) 10 800 (N) ;	[1]					
	(ii)	(work =) force × distance ; = 10800 × 100 = 1080 000 (J) ;	[2]					
	(iii)	(iii) (kinetic energy =) $\frac{1}{2}$ mv <sup>2</sup> ; = $\frac{1}{2} \times 1080 \times 0.2 \times 0.2 = 21.6$ (J);						
	(iv)	(average speed = ) distance/time ; =1000/4000 = 0.25(m/s) ;	[2]					
	<b>(c)</b> doe fuel	s not deplete fossil fuel reserves/carbon neutral/kerosene is a h	ydrocarbon [1]					
			[Total: 10]					
5	<b>(a)</b> sex	ual <u>and</u> asexual ; (both needed)	[1]					
	(b) (i)	(large) petals/large flowers/nectar guides/landing platform ;	[1]					
	(ii)							
		zygote produced ; (allow 'sex cell' or 'nucleus' instead of 'gamete')	[2]					
	(iii)	ovary;	[1]					
	(iv)	seeds ;	[1]					
	(c) (i)	increase growth of plants ; (plants need nitrates) to produce proteins ; proteins needed to produce new cells ;	[max 2]					
	(ii)	<ul> <li>Q has nitrogen-fixing bacteria in its roots ; provide plants with, nitrogen-containing compounds/ammonium ions ;</li> </ul>						
			ns ; [2] [Total: 10]					

	Page 5		5	Mark Scheme: Teachers' version IGCSE – October/November 2011	Syllabus 0654	Paper 22
6	(a)	(i)	89 ;		0034	[1]
				the component metals together ;		[1]
		(iii)	coins perio OR (che coins OR malle	ng/hard/low malleability ; s must not easily be damaged/must be easily ods/owtte ; mically) unreactive ; s must not easily corrode ; eable ; be shaped (for coin manufacture) ;	recognised over	long [max 2]
	(b)	(i)	tin o	xide + carbon $\rightarrow$ tin + carbon monoxide ;		[1]
		(ii)	carb joins	on ; with oxygen ;		[2]
	(c)	(i)	-	ative electrode ; d which conducts a current/contains mobile ions ;		[2]
		(ii)	(ator	n) loses electrons ;		[1]
		(iii)	15 ;			[1]
	(d)	(i)	sol/o	colloid ;		[1]
		(ii)		are) not transparent so light does not reach skin/( w sunscreen absorbs light/acts as a light filter)	sol) reflects light ;	[1]
						[Total: 13]
7	(a)	fror fror	n red	od ; (ignore 'ref to bleeding') blood cells ; moglobin ; on ;		[max 2]
	(b)	(i)	mak	bing warm ; ing new cells ; smitting nerve impulses ;		[max 2]
		(ii)	hom	eostasis ;		[1]
		(iii)		er lost in sweat ; o need to maintain water content of body ;		[max 2]

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(4	sup bloc cau	to coronary artery ; oplies heart muscle with, blood/oxygen ; ckage (in blood vessel) ; use of blockage (e.g. plaque, cholesterol, blood clot) ; art, muscle/cells, die/cannot contract, because of lack of oxygen ;	[max 3] <b>[Total: 10]</b>
8 (a	a) (i)	arrows go down ;	[1]
	(ii)	convection ;	[1]
	(iii)	cold air becomes more dense and sinks ;	[1]
(I	b) (i)	(density =) mass/volume ; = $7.4/8$ ; = $0.925 (g/cm^3)$ ;	[3]
	(ii)		[2]
(0	c) (i)	(resistance = ) voltage/current ; = $250/0.05 = 5000(\Omega)$ ;	[2]
	(ii)	$R_T = R_1 + R_2;$ = 5000 + 5000 = 10 000 (Ω);	[2]
			[Total: 12]
9 (a	a) (i)	A ; C ;	[2]
(I	b) (i)	photosynthesis ; respiration ;	[2]
	(ii)	glucose/carbohydrate/sugar/starch;	[1]
	(iii)	algae produce oxygen which coral uses/coral produces carbo which algae uses ;	n dioxide [1]

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(c) (i)		eased use of fossil fuel/example which implies this ; on dioxide produced when fuel burns ;	;	[2]
(ii)	mak	on dioxide dissolves in/reacts with (sea) water ; es water more acidic/less alkaline ; metal oxides are acidic ;		[max 2]
(iii)	e.g. it mo	ept any reasonable science based idea: calcium carbonate may react with more acidic wa ore difficult for coral to extract ions from sea / coral ive in more acidic water ;	•	
				[Total: 11]