UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

## MARK SCHEME for the October/November 2011 question paper

## for the guidance of teachers

## 0654 CO-ORDINATED SCIENCES

0654/21

Paper 2 (Core Theory), maximum raw mark 100

MMM. Hiremepapers.com

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

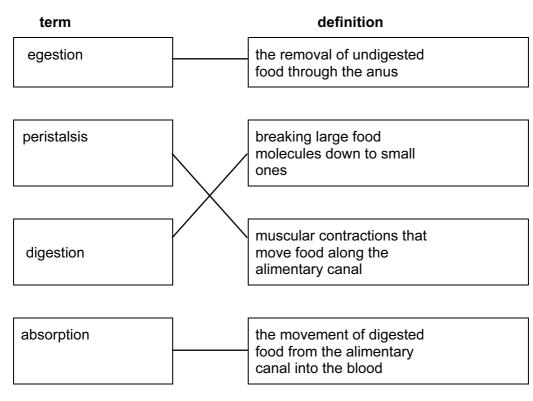
Cambridge is publishing the mark schemes for the October/November 2011 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.



Page 2		Mark Scheme: Teachers' version	Syllabus	Paper
		IGCSE – October/November 2011	0654	21
1 (a) (i)		tina ; btic nerve ;		[2]
(ii)		drawn starting outside the eye, passing through pup ends at retina ;	bil and lens ;	[2]
(iii)	labe	I to either cornea, lens or vitreous humour ;		[1]
(iv)		mpulse/electrical signal/action potentials ; ig nerve/along <b>Y</b> ;		[2]
(b) (i)	red	blood cell ;		[1]
(ii)	46 ;			[1]
(iii)	on n	es information; naking proteins ; to hereditary material ;		[max 2]
				[Total: 11]
2 (a) (i)		(no mark) on number of 6 ;		[1]
(ii)	11;			[1]
(b) (i)	cont	ains elements aluminium and oxygen ;		
(b) (i)		portions/number ratio $Al : O$ is 2:3/owtte ;		[2]
(ii)	ion (	(electrically) charged/ion protons $\neq$ electrons ;		[1]
(c) (i)		trolysis ; tive electrode ;		[2]
(ii)	•	tains) ions (which) must be mobile/if solid then ions duct electrical charge/make an electrolyte;	s cannot move ;	[2]
(iii)	→a	luminium + oxygen ;		[1]
				[Total: 10]

Pag	je 3	Mark Scheme: Teachers' version	Syllabus	Paper
		IGCSE – October/November 2011	0654	21
3 (a)	(i) cons	stant speed ;		[1]
(	(ii) dece	eleration/slowing down ;		[1]
	8/5 ; = 1.6(m/	's <sup>2</sup> );		[2]
		distance/time ; 5 = 6.4 (m/s) ;		[2]
		=) power × time ; 5 = 3000 (J) ;		[2]
	(water) c ref. to att fastest m (escape) point ;	sferred to (water) particles (from surroundings) ; hanges from liquid to gas ; raction between particles in the liquid ; noving particles escape ; at surface/ref. to process happening at temp energy of the rest of the particles reduced/heat rer		-
				[Total: 10]

## 4 (a)



all correct for 3 marks, 2 or 3 correct for 2 marks, 1 correct for 1 mark ;;;

Page 4	Mark Scheme: Teachers' version	Syllabus	Paper
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(b)

5

		[		1	
	enzyme	substrate	product		
	amylase	starch ;	maltose		
	protease ;	proteins	amino acids		
	lipase ;	fats/lipids ;	fatty acids and glycerol		
				[4]	
(i)	urea ;			[1]	
(ii)	kidneys ;			[1]	
(iii)			outside ;	[2]	
(iv)					
	glucose oxidised/glucose combined with oxygen ;				
		[Total: 13]			
(i)			[2]		
(ii)	electrical ; into light ; into heat ;			[3]	
(iii)	gas will not react with hot	filament/reduces oxidat	ion ;	[1]	
(i)	decreases ; to constant (minimum) va	lue ;		[2]	
(ii)	0.20(A);			[1]	
(i)	$R_T = R_1 + R_2;$ (= 1000 + 2000) $R = 3000(\Omega);$			[2]	
	(iii) (iv) (i) (ii) (ii) (ii) (ii)	amylase $protease ;$ $lipase ;$ (i) urea ;(ii) kidneys ;(iii) cells might take up water because concentration ins(iv) for energy ; respiration ; glucose oxidised/glucose for movement/other name(i) the power rating of bulb/h the voltage/potential diffe(ii) electrical ; into light ; into heat ;(iii) gas will not react with hot(i) decreases ; to constant (minimum) val(ii) 0.20 (A) ;(i) R <sub>T</sub> = R <sub>1</sub> + R <sub>2</sub> ; (= 1000 + 2000)	amylase starch ;   protease ; proteins   lipase ; fats/lipids ;   (i) urea ; (ii) kidneys ;   (iii) cells might take up water ; because concentration inside cells is greater than   (iv) for energy ; respiration ;   glucose oxidised/glucose combined with oxygen ;   for movement/other named use of energy ;   (i) the power rating of bulb/how much energy is trans   the voltage/potential difference that the bulb opera   (ii) electrical ;   into light ;   into light ;   into heat ;   (iii) gas will not react with hot filament/reduces oxidati   (i) decreases ;   to constant (minimum) value ;   (ii) 0.20 (A) ;   (ii) R <sub>1</sub> = R <sub>1</sub> + R <sub>2</sub> ;   (ii) R <sub>1</sub> = R <sub>1</sub> + R <sub>2</sub> ;   (iii) 0.20 (A) ;	amylase starch ; maltose   protease ; proteins amino acids   lipase ; fats/lipids ; fatty acids and glycerol   (i) urea ; (ii) kidneys ; (iii) cells might take up water ; because concentration inside cells is greater than outside ;   (iv) for energy ; respiration ; glucose oxidised/glucose combined with oxygen ; for movement/other named use of energy ;   (i) the power rating of bulb/how much energy is transferred per second ; the voltage/potential difference that the bulb operates at ;   (ii) electrical ; into light ; into heat ;   (iii) gas will not react with hot filament/reduces oxidation ;   (i) decreases ; to constant (minimum) value ;   (ii) 0.20(A) ;   (ii) R <sub>T</sub> = R <sub>1</sub> + R <sub>2</sub> ; (= 1000 + 2000)	

(ii) cross sectional area/thickness ; temperature of wire ; [2]

[Total: 13]

	Page 5		6	Mark Scheme: Teachers' version	Syllabus	Paper
				IGCSE – October/November 2011	0654	21
6	(a)	(i)	( <b>B</b> )	low electrical conductivity ;		[1]
		(ii)	( <b>C</b> ) ł	high density and (high) electrical conductivity ;		[1]
	(b)	(i)	all th	nree metals must be melted together ;		[1]
		(ii)	sold	er maintains the electrical connection/owtte ;		[1]
	(c)	con suit indi	itaine able catioi	electrode ; r plus liquid ; named electrolyte (into which electrodes are placed) n that second electrode is a different metal ; n that pd is produced e.g. value on voltmeter ;	);	[max 3]
	(d)	(i)		oon dioxide ; er (vapour) ;		[2]
		(ii)	carb	oon monoxide/NOx or specific example/hydrocarbo	ns ;	[1]
						[Total: 10]
7	(a)	(i)	(frec	quency =) 0.5 (Hz) ;		[1]
		(ii)		ter/mass ; els/moves/goes ;		[2]
		(iii)	sour	nd/ultra sound/infrasound ;		[1]
	(b)	(i)	500	(N) ;		[1]
		(ii)		rk done =) force × distance ; )0 × 10 = 5000(J) ;		[2]
		(iii)		etic energy =) ½ mv² ; × 50 × 12 × 12 = 3600 (J) ;		[2]
	(c)	rad	iation	ı;		[1]
						[Total: 10]

	Page 6		Mark Scheme: Teachers' version	Syllabus	Paper
			IGCSE – October/November 2011	0654	21
8	(a) (i)	bino	omial ;		[1]
	(ii)	idea	that it is the same all over the world/in every lang	guage ;	[1]
	(b) (i)		snakes bamboo rats ow golden lion tamarins	/ls	
		plan	trees / nectar / fruit (allow if in separate boxes) Its and tamarins connection correct ;		
			nree predators connection correct ; rrows in right direction ;		[3]
	(ii)	circle	e round tree, nectar or fruit ;		[1]
	(c) (i)	furth	er faeces further from tree ; nest distance from tree is 400 m ; res quoted, e.g. 31% of faeces deposited within 50	)m of tree ;	[max 2]
	(ii)	less facto	es provide nutrients for, young plants/seedlings ( competition (for seedlings) away from parent tree or competed for – water/light ;		
		nelp	to colonise new areas ;		[max 2]
					[Total: 10]

	Page 7		Mark Scheme: Teachers' version	Syllabus	Paper
			IGCSE – October/November 2011	0654	21
9	(a) (i)		nent made of one type of atom/found in Periodic Ta pound made of two or more different elements ;	ble ;	[2]
	(ii)	•	np red) litmus paper/(red) litmus solution/full range s blue ;	/Universal Indicato	or ; [2]
	(b) (i)	nitro	gen inert/difficult to break $N_2$ molecule ;		[1]
	(ii)	nitric	c acid ;		[1]
	(iii)		on, hydrogen, oxygen ;; hree – 2 marks, only two correct – 1 mark)		[2]
	(iv)	-	link/react together ; rm a (long) chain ;		[2]
	(c) (i)		eases rate of reaction ; consumed/used up/can be regenerated ;		[2]
	(ii)		ould react/corrode/sodium does not have cataly a transition metal ;	tic properties/sodi	um [1]
					[Total: 13]