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#### **CAMBRIDGE INTERNATIONAL EXAMINATIONS**

**International General Certificate of Secondary Education** 

## MARK SCHEME for the October/November 2013 series

# 0654 CO-ORDINATED SCIENCES

**0654/53** Paper 5 (Practical), maximum raw mark 45

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2013 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.



Page 2		2	Mark Scheme	Syllabus	Paper	
				IGCSE – October/November 2013	0654	53
1	(a)	(i)	barle	ey grains drawn in both dishes ;		[1]
		<ul><li>(ii) drawings of both dishes;</li><li>dish A shows brown/orange/yellow and blue/black areas labelled</li><li>AND</li></ul>		rol		
			no b	rown in dish <b>B</b> ;		[2]
	(iii) brown/orange/yellow colour around where the barley grains were; (allow no starch where grains were)		[1]			
	1	(iv) (enzyme from the) barley grains breaking down/digesting the starch; (allow area below grains no longer contains starch)			[1]	
		(v)	cont	rol/shows that breakdown depends on living barley	grains;	[1]
	(b)	(i)		our sections of the table filled in ; row shows Benedict's solution blue ;		[2]
		(ii)		<b>B</b> : (reducing) sugar present/not present; (concervation)	nclusion must m	natch
			beca	ause starch digested to sugar/sugar absorbed anation; (reason must match conclusion)	by seed/other of	good [2]
				<b>D</b> : (reducing) sugar not present ; ause seeds are dead / starch not digested ;		[2]
	(c)	imp	roved	d reliability/because one seed might not be active/o	owtte;	[1]
	(d)	(i)	large	er brown areas/less starch present ;		[1]
		(ii)	sma	ller brown areas/more starch present;		[1]

[Total: 15]

	Page 3		Mark Scheme	Syllabus	Paper
			IGCSE – October/November 2013	0654	53
2	(a) (i	) rea	ding for $x$ when $d = 55 \mathrm{cm}$ to nearest mm;		[1]
	(ii	-	e reading on either side of mass and find the mean divide by 2 to find centre mark;	value/measure r	nass [1]
	(iii	,	nplete set of <i>x</i> values ; alues increasing down the table ;		[2]
	(b) (i	suit at le	s labelled with units ; able choice of scales (points use at least 8 cm × 8 cr east 4 points correct to half a small square ; d best fit straight line judgement ;	n of grid) ;	[4]
	(ii	AN at le	cation on graph of how data obtained  Deast half of line used; rect calculation from triangle method using data gnificant figures);	from graph (at	least [2]
	`´ fiç	<ul><li>(c) correct calculation of m (from candidate's gradient value, to 2/3 significant figures)</li><li>AND</li></ul>		icant	
	CC	orrect	rounding required;		[1]
	(d) (i	•	n values present, and realistic ; n values to nearest millimetre ;		[2]
	(ii	2 si	rect calculation of density (from candidate's v gnificant figures) ; uracy mark: value within ± 0.1 of Supervisor's value		least [2]

[Total: 15]

Page 4	Mark Scheme	Syllabus	Paper
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## 3 Table 3.1

	test	observation	conclusion	
(a) (i)	dilute nitric acid	no reaction/nothing/paler solution;	no carbonate/no CO <sub>3</sub> <sup>2-</sup> ;	
(ii)	barium chloride solution	ppt of stated colour ;	sulfate / SO <sub>4</sub> <sup>2-</sup> ;	
(iii)	silver nitrate solution	white ppt;	chloride/Cl <sup>-</sup> ;	

[6]

### Table 3.2

	test	observation	conclusion		
(b) (i)	sodium hydroxide solution	brown / orange / red-brown / yellow-brown <b>AND</b> ppt/residue; colourless filtrate;	iron(III) / Fe <sup>3+</sup> ;		
(ii)	ammonia solution	brown / orange / red-brown / yellow-brown <b>AND</b> ppt residue; dark blue filtrate;	iron(III)/Fe <sup>3+</sup> ; copper(II)/Cu <sup>2+</sup> ;		
(iii)	sodium carbonate solution	brown ppt ;	, sappo.(11), sa ,		

[8]

(c) iron(III) chloride AND copper(II) sulfate / iron(III) sulfate AND copper(II) chloride; (allow any three or all four compounds but not a list of the ions) [1]

[Total: 15]